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## SECTION 5.1 THE CELL CYCLE

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1. kinetic theory Of matter 2. plasma 3. crystals 4. solid rzoáamorphous material 6. steam \_e\_ 7. thermal expansion 8. liquid 10. ice use the words in the box to fill in the blanks. Column II water vapor b. state of matter with no definite shape but with definite volume c. solid which is not made of crystals d.

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in terms of particles in . 2. The kinetic-molecular theory makes the following assumptions. a. In a sample of a gas, the volume of the gas particles themselves is very

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Chapter 10: Section 1: The Kinetic-Molecular - Section 1: The Kinetic-Molecular Theory of Matter kinetic-molecular theory. Study Everywhere! 2015 Quizlet Inc. Follow @quizlet on Twitter. Presentation " Section 1. Kinetic Theory: How - Section 1. Kinetic Theory: How particles in matter behave 3 Basic Assumptions of the Kinetic Theory 1. All matter is composed of small particles (atoms,

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rscott49. Terms in this set (14) Kinetic Theory. Explains how the particles in gases behave. Plasma. Gas like mixture of charged particles. Crystals. Particles arranged in repeating geometric ...

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